

Redetermination of  $\alpha$ -ammonio- $\alpha$ -methylpropionateDaniel E. Lynch<sup>a\*</sup> and Ian McClenaghan<sup>b</sup><sup>a</sup>School of Science and the Environment, Coventry University, Coventry CV1 5FB, England, and <sup>b</sup>Key Organics Ltd, Highfield Industrial Estate, Camelford, Cornwall PL32 9QZ, EnglandCorrespondence e-mail:  
apx106@coventry.ac.uk

## Key indicators

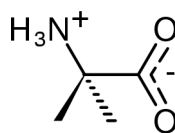
Single-crystal X-ray study  
 $T = 150$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å  
 $R$  factor = 0.045  
 $wR$  factor = 0.130  
Data-to-parameter ratio = 15.3For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The redetermined structure of the title compound,  $\text{C}_4\text{H}_9\text{NO}_2$ , comprises a simple amino acid derivative arranged in an extensive hydrogen-bonded network, with the three hydrogen-bond donor elements (N–H) associating with the two hydrogen-bond acceptor elements (carboxylate O atoms). The addition of one C–H···O close contact results in both carboxylate O atoms being involved in three-centre hydrogen-bonding associations.

Received 23 May 2002  
Accepted 29 May 2002  
Online 8 June 2002

## Comment

The redetermined structure of the title compound, (I), comprises a simple amino acid derivative arranged in an extensive hydrogen-bonded network, with the three hydrogen-bond donor elements (N–H) associating with the two hydrogen-bond acceptor elements (carboxylate O atoms). The addition of one C–H···O close contact results in both carboxylate O atoms being involved in three-centre hydrogen-bonding associations. The structure of (I) was initially determined in 1952 (Hirokawa *et al.*, 1952), using photographic techniques; the  $R$  value was 0.199. With such poor data it was impossible to confirm the zwitterionic nature of the compound, which is revealed in this current study. The title compound has been used as both a ligand and organic counter-ion in 19 other reported crystal structures.



(I)

## Experimental

The title compound was obtained from Key Organics Ltd and crystals were grown from an ethanol solution.

## Crystal data

$\text{C}_4\text{H}_9\text{NO}_2$   
 $M_r = 103.12$   
Monoclinic,  $C2/c$   
 $a = 10.6273$  (8) Å  
 $b = 9.0102$  (8) Å  
 $c = 11.3370$  (7) Å  
 $\beta = 93.749$  (6)°  
 $V = 1083.2$  (1) Å<sup>3</sup>  
 $Z = 8$

$D_x = 1.265$  Mg m<sup>-3</sup>  
Mo  $K\alpha$  radiation  
Cell parameters from 2205 reflections  
 $\theta = 2.9$ – $27.5$ °  
 $\mu = 0.10$  mm<sup>-1</sup>  
 $T = 150$  (2) K  
Block, colourless  
 $0.30 \times 0.10 \times 0.08$  mm

## Data collection

Bruker–Nonius KappaCCD area-detector diffractometer  
 $\varphi$  and  $\omega$  scans  
 Absorption correction: multi-scan (SORTAV; Blessing, 1995)  
 $T_{\min} = 0.970$ ,  $T_{\max} = 0.992$   
 3853 measured reflections

1208 independent reflections  
 755 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.070$   
 $\theta_{\text{max}} = 27.4^\circ$   
 $h = -13 \rightarrow 11$   
 $k = -10 \rightarrow 11$   
 $l = -14 \rightarrow 13$

## Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.045$   
 $wR(F^2) = 0.130$   
 $S = 1.00$   
 1208 reflections  
 79 parameters  
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0628P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.17 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.19 \text{ e } \text{\AA}^{-3}$   
 Extinction correction: SHELXL97  
 Extinction coefficient: 0.013 (3)

Table 1

Hydrogen-bonding geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$N1-H1 \cdots O2^i$	1.04 (2)	1.84 (2)	2.8645 (18)	167 (2)
$N1-H2 \cdots O1^{ii}$	0.96 (2)	1.88 (2)	2.828 (2)	170 (2)
$N1-H3 \cdots O2^{iii}$	0.97 (3)	1.90 (3)	2.841 (2)	163 (2)
$C3-H31 \cdots O1^i$	0.98	2.42	3.326 (2)	153

Symmetry codes: (i)  $x, -y, \frac{1}{2} + z$ ; (ii)  $1 - x, y, \frac{1}{2} - z$ ; (iii)  $\frac{3}{2} - x, \frac{1}{2} + y, \frac{1}{2} - z$ .

All methyl H atoms were included in the refinement, at calculated positions, as riding models with C–H distances set to 0.98  $\text{\AA}$ , whereas the ammonium H atoms were located in difference syntheses and both positional and displacement parameters were refined.

Data collection: DENZO (Otwinowski & Minor, 1997) and COLLECT (Hooft, 1998); cell refinement: DENZO and COLLECT; data reduction: DENZO and COLLECT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: PLATON97 (Spek, 1997); software used to prepare material for publication: SHELXL97.

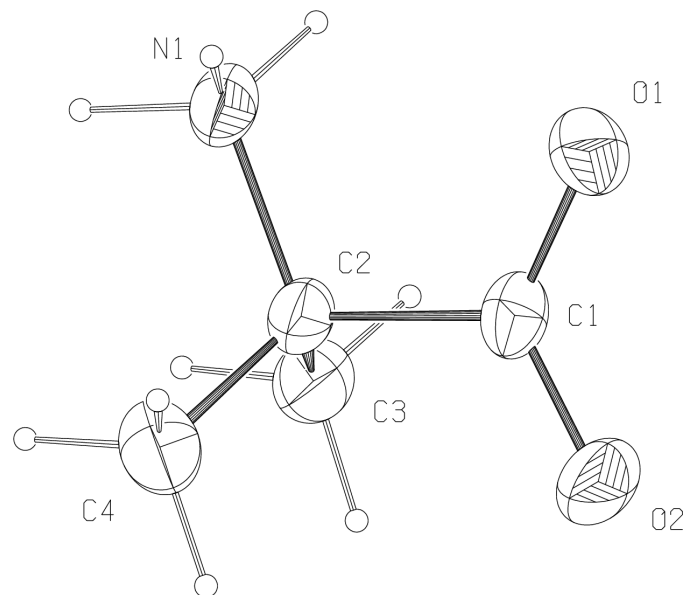


Figure 1

The molecular configuration and atom-numbering scheme for (I), showing 50% probability ellipsoids.

The authors thank the EPSRC National Crystallography Service (Southampton).

## References

- Blessing, R. H. (1995). *Acta Cryst.* **A51**, 33–37.  
 Hirokawa, S., Kuribayashi, S. & Nitta, I. (1952). *Bull. Chem. Soc. Jpn.*, **25**, 192–195.  
 Hooft, R. (1998). *COLLECT*. Nonius BV, Delft, The Netherlands.  
 Otwinowski, Z. & Minor, W. (1997). *Methods in Enzymology*, Vol. 276, *Macromolecular Crystallography*, Part A, edited by C. W. Carter Jr and R. M. Sweet, pp. 307–326. New York: Academic Press.  
 Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.  
 Spek, A. L. (1997). *PLATON97*. Version of May 1997. University of Utrecht, The Netherlands.